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α -Hydroxymethylation of γ - and δ -Lactones: a New Synthesis of α -Methylene- γ -butyrolactones

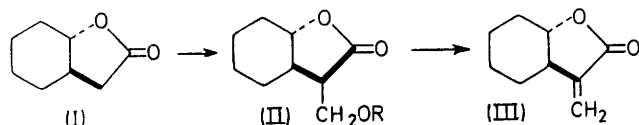
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Summary The direct high yield α -hydroxymethylation of the *trans*-bicyclic lactone (I) and δ -valerolactone, and an efficient three-step synthesis of the *trans*- α -methylene- γ -butyrolactone (III) are described.

THE α -methylene- γ -butyrolactone structural unit is found in many naturally occurring cytotoxic sesquiterpene lactones which have been isolated during the search for antitumor agents from plant sources,¹ and its synthesis has recently

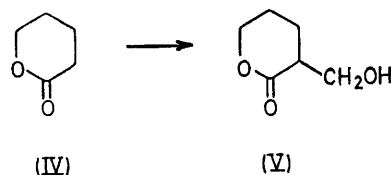
δ units (*e.g.* vernolepin)³ we required a procedure for the introduction of the α -methylene unit from lactone precursors, and we report the direct conversion of a γ -butyrolactone into an α -hydroxymethyl- γ -butyrolactone [*e.g.* (I) \rightarrow (IIa)] with lithium di-isopropylamide [*e.g.* (I) \rightarrow (IIa)] with lithium di-isopropylamide to form the lactone enolate⁴ which is efficiently trapped with formaldehyde. To a solution of lithium di-isopropylamide



a, R = H
b, R = MeSO₂
c, R = *p*-MeC₆H₄SO₂

received much attention.² However, none of the many synthetic schemes permits access *via* direct α -hydroxymethylation of the γ -butyrolactone system, despite the ready availability of the appropriate lactone precursors. The method described here is applicable to δ -lactones; other methods are not applicable to the six-membered ring lactones.^{2c-e}

In connection with other work on the total syntheses of sesquiterpenes possessing both the α -methylene γ and



(1.2 equiv.) in anhydrous tetrahydrofuran (THF) at -78° was slowly added a solution of lactone (I) (1 equiv.) in THF. After 30 min the mixture was warmed to -20° and gaseous formaldehyde in nitrogen was passed in. After an additional 30 min, the reaction was quenched by the addition of 10% HCl. Work-up afforded the butyrolactone (IIa) (>95%),[†] which was converted into the mesylate (IIb) (MeSO₂Cl-pyridine) and thence (refluxing pyridine) into the known *trans*- α -methylene- γ -butyrolactone (III) [ν_{\max} (film) 1771 and 1674 cm⁻¹; δ (CDCl₃) 6.04 (d, *J* 3 Hz, C=CH₂), 5.35 (d, *J* 3 Hz, C=CH₂), 3.45–3.95 (br, CHO) p.p.m.]. The overall yield from (I) to (III) was *ca.* 80%. In addition, conversion of (IIa) into tosylate (IIc), m.p. 112–112.5°,

[†] Satisfactory analytical and/or spectral data were obtained for all compounds.

followed by refluxing of (IIc) in pyridine afforded (III) in *ca.* 75% overall yield from (IIa).

In view of the lack of methods for the construction of the α -methylene- δ -valerolactone system, we applied the direct hydroxymethylation procedure to δ -valerolactone (IV). Treatment of (IV) with lithium di-isopropylamide as already described, followed by treatment with formalde-

hyde at -20° afforded the α -hydroxymethyl- δ -valerolactone (V) (> 95%).

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⁴ A report describing the alkylation of lactone enolates with methyl iodide has recently appeared (G. H. Posner and G. L. Loomis, *J.C.S. Chem. Comm.*, 1972, 892.)